electrical properties of PbMoO₄ single crystals



T.M.Bochkova¹*, M.P. Trubitsyn¹, M.D. Volnianskii¹, D.M. Volnyanskii²

¹ Institute for Energy Efficient Technologies and Materials Sciences, Oles Honchar Dnipro National University, prosp. Gagarina 72, Dnipro, 49010, Ukraine;

² Ukrainian State University of Science and Technologies, Lazaryan str. 2, 49010 Dnipro, Ukraine

E-mail: tbochkova@meta.ua

Introduction

 $PbMoO_4$ crystals are widely used in modern electronics as multifunctional materials. The main areas of applications are acousto-optics, laser technology and cryogenic scintillation systems for detecting rare events in physics of elementary particles. These applications require large-sized crystals of good quality. However, PbMoO₄ crystals grown from a melt on a seed, as a rule, contain characteristic structural imperfections. Among them there are nanometer-sized clusters, including intrinsic point defects and impurities, photoinduced complexes based on anionic MoO₄ groups with trapped photoelectrons, inclusions of foreign phases. Such defects have a strong effect on the electrical and optical properties of crystals. In this work, we study the effect of variations of the initial mixture composition and UV irradiation on polarization and charge transfer processes in PbMoO₄ single crystals.

Experimental details

The single crystals of PbMoO₄ were grown from the melt by conventional Czochralski technique in air by using platinum crucibles. The charge was prepared by solid phase synthesis at 925-975 K for 2 hours from MoO₃ and PbO of α - or β - modifications of "high purity" grade. The reagents were taken both in a stoichiometric ratio and with excess of 0,5 mol% of MoO_3 . The crystals grown in the direction deviated by 30° from *a*-axis in (001) plane were free from gas bubbles and cracks. The main planes of the samples with dimensions $5 \times 5 \times 1$ mm³ were cut perpendicular to the growth axis. The main faces of the samples were irradiated using light-emitting diode with radiation wavelength λ =365–370 nm for 30 – 60 minutes. Then platinum electrodes were deposited by cathode sputtering. Permittivity ε and conductivity σ were measured in AC field by the bridge method at fixed frequency (f=1 kHz) in the temperature interval 290–700 K.

Results and discussion





Fig.1. PbO can exist in two polymorphic forms. This is an α -modification (lead litharge), which has a red color and belongs to the tetragonal syngony (symmetry space group P 4/nmm, Z=4, unit cell parameters: a=0.3976 nm; c=0.5023 nm, figure on the left). And β -modification (massicot), the crystals of which are yellow in color and are characterized by rhombic symmetry (space group *P bcm*, Z=2, unit cell parameters: *a*=0.5489 nm; *b*=0.4755 nm; *c*=0.5891 nm, figure on the right).



Within the framework of the hopping conductivity model, we suppose that the observed changes in electrical properties are associated with a redistribution of the density of localized states in the bandgap of PbMoO₄ crystals, caused by the appearance of the structural and the photoinduced nanometer-sized defects in the crystal lattice.







Fig. 2. The temperature dependences of AC conductivity of PbMoO₄ crystals before (left) and after UV irradiation (right): 1 – crystals were grown using β – PbO; 2 - crystals were grown using α – PbO.



Fig.4. The temperature dependence of AC conductivity (f=1 kHz) of pre-illuminated with UV light PbMoO₄ crystals: the crystals grown using β -modification (a) and α -modification of PbO (b) (1 – the charge of stoichiometric composition, 2 - the charge with 0.5 mol% MoO₃ excess).

Fig. 3. Permittivity dependencies $\varepsilon(T)$ for PbMoO₄ crystals of stoichiometric composition: 1 – crystals were grown using β – PbO; 2 - crystals were grown using α – PbO. The measurements were carried out after irradiation of the samples with UV light.



Fig.5. Permittivity dependencies $\varepsilon(T)$ for PbMoO₄ crystals grown using β -PbO (left) and α – PbO (right): 1 - stoichiometric composition, 2 - composition with excess of 0,5 mol % MoO₃. The measurements were carried out after irradiation of the samples with UV light.

